Structural studies on bioactive compounds. Part 37.1 Suzuki coupling of diaminopyrimidines: a new synthesis of the antimalarial drug pyrimethamine[†]

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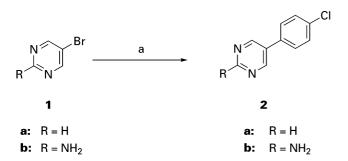
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Suzuki reactions have been used successfully to effect cross-coupling of 5-halopyrimidines with 4-chloroben-zeneboronic acid and 2,4-diamino-5-(4-chloro-3-halo)-6-ethylpyrimidines with 4-methoxybenzeneboronic acid. The antimalarial drug pyrimethamine has been prepared by coupling 2,4-diamino-6-ethyl-5-iodopyrimidine with 4-chlorobenzeneboronic acid.

Keywords: bioactive compounds, diaminopyrimidines, pyrimethamine

Early efforts to exploit the Suzuki cross-coupling procedure² to attach aryl residues to halopyrimidines showed that neutral pyrimidine substrates coupled efficiently.³ However, we have shown that amphoteric 2-amino-5-halopyrimidin-4(3*H*)-ones couple with difficulty even with normally reactive boronic acids, and dehalogenation is often the predominant result.⁴ Presumably, under the basic conditions normally required, the aminohalo-pyrimidinone is in the anionic (electron-rich) state where cross-coupling is disfavoured. Conversely, it would be expected that aminohalo-pyrimidines should couple readily. We have demonstrated this principle and achieved a new synthesis of the antimalarial drug pyrimethamine (6).

In pilot experiments, 5-bromopyrimidine (1a) and 2-amino-5-bromopyrimidine (1b) were coupled with 4-chlorobenzeneboronic acid in 1,2-dimethoxyethane (DME) in the presence of potassium phosphate and the catalyst Pd(OAc)₂(dppf)₂, prepared *in situ* from 1,1'-(diphenylphosphino)ferrocene (dppf) and palladium acetate. 5-(4-Chlorophenyl)pyrimidine (2a) was isolated in 51% yield (Scheme 1) and the higher yield of 2b (74%) from the aminobromo-pyrimidines (1b) encouraged the belief that diaminohalo-pyrimidines would be suitable substrates for cross-coupling reactions. However, despite a range of solvents, bases and catalysts being studied no 5-(2-chlorophenyl)pyrimidines were isolated when 2-chlorobenzeneboronic acid was employed as coupling partner in reactions with 1a and 1b.



Scheme 1
Reagents and conditions: a, Pd(OAc)₂(dppf)₂, 4-chloroben-zeneboronic acid, K₂HPO₄, DME, reflux under N₂

Our new route to pyrimethamine (6) started with the aminopyrimidinone (3), prepared (97%) from methyl propionylacetate and guanidine carbonate.⁵ This was converted to the aminochloro-pyrimidine (4) (90%) with phosphorus oxychloride and thence to the diaminoiodo-pyrimidine (5) by (successive) amination and iodination reactions in overall 50% yield (Scheme 2). Coupling of 5 with 4-chlorobenzeneboronic acid using the (above) conditions afforded 6 in 76% recrystallised yield. No coupled products were detected when attempts were made to react 5 with 2-chlorobenzeneboronic acid or the *N*,*N*'-diacetyl-derivative of 5 with 4-chlorobenzeneboronic acid under the standard conditions employed.

The cross-coupling route represents a practical improvement over the historic synthesis of pyrimethamine by Russell and Hitchins⁶ which involves interaction of a 3-carbon synthon and a guanidinium salt and employs diazomethane at one stage. A possible alternative route to pyrimethamine from 4 can be envisaged from (successive) methanolysis and bromination to form the bromomethoxy-pyrimidine (7), followed by cross-coupling with 4-chlorobenzeneboronic acid. The product (8) was only isolated in impure form (50%) after repeated chromatographic fractionation to remove starting material (7) and its debrominated analogue. Accordingly, 8 was not subjected to ammonolysis which would have led to pyrimethamine (6).

A further dimension can be added to the cross-coupling reactions of pyrimethamine starting from aminopyrimethamine (**9a**). Diazotisation of **9a** in tetrafluoroboric acid is known to yield the stable diazonium tetrafluoroborate salt (**9b**). We have now shown that Sandmeyer reactions on diazonium salts from **9a** afforded the new bromo- (**9c**) and iodopyrimethamine (**9d**) in 74 and 85% yields: these substrates (**9b–d**) were coupled with 4-methoxybenzeneboronic acid to furnish the substituted biphenyl (**10**) in 47, 47 and 20% yields, respectively. Presumably, the chlorophenyl groups of both pyrimethamine (**6**) and the 3'-(4-methoxyphenyl)-pyrimethamine (**10**) would also undergo further cross-couplings, but this has not been explored.

Experimental

Melting points were obtained on a Gallenkamp melting point apparatus and are uncorrected. IR spectra were obtained on a Mattson 2020 Galaxy Series FT-IR spectrophotometer. All NMR spectra were acquired on a Bruker ARX 250 instrument. Chemical shifts are reported in δ units and referenced to tetramethylsilane as internal standard; coupling constants (J) are in Hz. New compounds were characterised by 1 H and 13 C NMR and HRMS because of the propensity of aminopyrimidines to undergo partial solvation resulting in variable elemental analyses.⁸

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 $[\]dagger$ This is a Short Paper, there is therefore no corresponding material in *J Chem. Research (M)*.

$$H_2N$$
 H_2N
 H_2N

Scheme 2

Reagents and conditions: a, POCl₃, reflux; b, 35% aq NH₃, sealed tube; c, l₂ in aq NaOH; d, see a (Scheme 1); e, NaOMe, in MeOH, reflux; f, Br₂ in AcOH, 70°C; g, see a (Scheme 1)

$$\begin{array}{c|c}
NH_2 & C \\
N & Et
\end{array}$$
9

a:
$$R = NH_2$$

a b: $R + N_2 + BF_4$

b c: $R = Br$

c d: $R = I$

d

H₂N

NH₂

OMe

Scheme 3

Reagents and conditions: a, NaNO₂ in 50 % aq HBF₄ at 0°C; b, NaNO₂ in HBr at 0°C, then Cu(I)Br; c, NaNO₂ in H2SO₄ at 0°C, then KI; d, Pd(OAc)₂(dppf)₂, 4-methoxybenzeneboronic acid, K₂HPO₄, DME, reflux under N₂

2-Amino-5-bromo-6-ethyl-4-methoxypyrimidine (7): Prepared (98%), from the bromination of 2-amino-6-ethyl-4-methoxypyrimidine in acetic acid at 70°C, this *pyrimidine* had m.p. 93–95°C; v_{max} (KBr/cm⁻¹) 3459, 3162, 1632, 1561, 1481, 1447, 1379, 1044, 789; δ_{H} (2H₆]DMSO) 1.14 (3 H, t, *J* 7.5, CH₃), 2.63 (2 H, q, *J* 7.5, CH₂), 3.87 (3 H, s, OCH₃), 6.82 (2 H, br s, NH₂); δ_{C} ([²H₆]DMSO) 12.1 (CH₃), 29.6 (CH₂), 54.3 (OCH₃), 90.4 (C), 161.7 (C), 165.3 (C), 169.8 (C).

2,4-Diamino-5-(3-bromo-4-chlorophenyl)-6-ethylpyrimidine (9c). Copper (II) sulfate pentahydrate (0.45 g, 1.8 mmol) and sodium bromide (0.91 g, 1.9 mmol) in water (1.5 ml) were heated to 45°C and sodium metabisulfite (0.12 g, 0.63 mmol) in water (1 ml) was added in portions. Additional sodium metabisulfite was added to the hot solution until the blue colour was discharged. The supernatant was removed and the precipitate was washed by decantation (water) and the solid dissolved in 48% hydrobromic acid (0.5 ml) to give a solution of copper (I) bromide. 2,4-Diamino-5-(3-amino-4-chlorophenyl)-6-ethylpyrimidine (9a, 1.02 g, 3.87 mmol)⁷ in 48% hydro bromic acid (1.4 ml) at 0°C was diazotised by the addition of sodium nitrite (0.27 g, 3.87 mmol) in water (0.6 ml). The diazonium salt solution was added in portions to boiling copper (I) bromide solution (above). Addition of water followed by basification with concentrated aqueous ammonia/ice liberated the free base (9c; bromopyrimethamine) (74%), m.p. 222–223°C; ν_{max} (KBr/cm⁻¹) 3437, 3310, 3163, 1626, 1561, 1431, 1121, 1020, 812; δ_{H} ([$^{2}H_{6}$]DMSO) 1.07 (3 H, t, J 7.5, CH₃), 2.21 (2 H, q, J 7.5, CH₂), 5.83 (2 H, br s, NH₂), 6.04 (2 H, br s, NH₂), 7.31 (1 H, dd, J 2.0, 8.25, H-6'), 7.65 (1 H, d, J 2.0, H-2'), 7.76 (1 H, d, J 8.25, H-5'); δ_{C} ([²H₆]DMSO) 13.2 (CH₃), 27.5 (CH₂), 104.5 (C), 122.0 (C), 131.0 (CH), 132.1 (C), 135.9 (CH), 137.3 (C), 162.1 (C), 163.2 (C), 166.4 (C); m/z (HRMS-EI) 325.9925 (M+). C₁₂H₁₂BrClN₄ requires 325.9934

2,4-Diamino-5-(4-chloro-3-iodophenyl)-6-ethylpyrimidine (9d): Prepared from 9a by diazotisation in 2M-sulfuric acid, a Sandmeyer reaction with aqueous potassium iodide, followed by basification (aqueous concentrated ammonia), the free base (9d; iodopyrimethamine) (85%) was crystallised from aqueous ethanol, m.p. 205°C; ν_{max} (KBr/cm⁻¹) 3459, 3312, 3163, 1555, 1431; $δ_{\rm H}$ ([²H₆]DMSO) 1.12 (3 H, t, J 7.5, CH₃), 2.25 (2 H, q, J 7.5, CH₂), 5.95 (2 H, br s, NH₂), 6.15 (2 H, br s, NH₂), 7.36 (1 H, dd, J 2.0, 8.25, H-6'), 7.76 (1 H, d, J 8.25, H-5'), 7.86 (1 H, d, J 2.0, H-2'); $δ_{\rm C}$ ([²H₆]DMSO) 13.1 (CH₃), 27.1 (CH₂), 99.6 (C), 104.6 (C), 129.9 (CH), 132.6 (CH), 136.3 (C), 136.4 (C), 142.1 (CH), 161.3 (C), 162.4 (C), 165.0 (C); m/z (HRMS-EI) 373.9784 (M+). C_{12} H₁₂CIIN₄ requires 373.9795.

General method for Suzuki cross-coupling reactions: Palladium acetate (6.7 mg, 0.03 mmol) and 1,1'-(diphenylphosphino)ferrocene (22 mg, 0.04 mmol) were heated in degassed 1,2-dimethoxyethane (4 ml) for 15 min under nitrogen. To the cooled solution was added the halo-pyrimidine (1.56 mmol), the substituted benzeneboronic acid 2.03 mmol), potassium phosphate (0.662 g, 3.12 mmol), 1,2-dimethoxyethane (6 ml) and water (3 ml). The mixture was heated at reflux for 24 h under nitrogen. After cooling, the reaction mixture was diluted with water (10 ml), extracted with chloroform (3 × 20 ml), washed with brine (20 ml) and water (20 ml), and the organic layer was evaporated. The residue was crystallised as appropriate (below).

The following cross-coupled products were prepared: 5-(4chlorophenyl)pyrimidine (2a), from 1a (51%), from ethanol, m.p. 154–156°C (lit., 9 m.p. 154–155°C); v_{max} (KBr/cm⁻¹) 1555, 1418, 1354; $\delta_{\rm H}$ ([²H₆]DMSO) 7.63 (2 H, dt, J 2.5, 8.5, H-2′,6′), 7.89 (2 H, J 2.5, 8.5, H-3',5'), 9.18 (2 H, s, H-4,6), 9.23 (1 H, s, H-2); $\delta_{\rm C}$ $([^{2}H_{6}]DMSO)$ 128.9 $(2 \times CH)$, 129.4 $(2 \times CH)$, 132.2 (C), 132.8 (C), 134.1 (C), 154.8 (2 \times CH); 2-amino-5-(4-chlorophenyl)pyrimidine (2b), from 1b (74%), from ethanol, m.p. 193–195°C (lit., 10 m.p. 198°C); v_{max} (KBr/cm⁻¹) 3298, 3163, 1665, 1597, 1491, 826; δ_{H} ([²H₆]DMSO) 6.87 (2 H, br s, NH₂), 7.49 (2 H, d, J 7.5, H-2',6'), 7.67 $(2 \text{ H}, J 7.5, \text{H}-3', 5'), 8.60 (2 \text{ H}, \text{ s}, \text{H}-4,6); \delta_{\text{C}}([^{2}\text{H}_{6}]\text{DMSO}) 121.0 (C),$ 127.1 (2 × CH), 129.1 (2 × CH), 131.8 (C), 134.4 (C), 156.1 (2 × CH), 163.2 (C); 2,4-diamino-5-(4-chloro-phenyl)-6-ethylpyrimidine (6; pyrimethamine), from 5 (76%), m.p. 240-241°C, identical (m.p., IR, ¹H and ¹³C NMR) to an authentic sample purchased from Sigma 2-amino-5-(4-chlorophenyl)-6-ethyl-4-Company; methoxypyrimidine (8), from 7 (50%), after fractionation of the product by flash chromatography in ethyl acetate-hexane, 1:1), m.p. 172–174°C; v_{max} (KBr/cm⁻¹) 3472, 3297, 1636, 1580, 1450, 1370, 1260; $\delta_{\rm H}$ ([2 H₆]DMSO) 1.02 (3 H, t, J 7.5, CH₃), 2.27 (2 H, q, J 7.5, CH₂), 3.73 (3 H, s, OCH₃), 6.60 (2 H, br s, NH₂), 7.23 (2 H, d, J 8.5, H-2',6'), 7.45 (2 H, J 8.5, H-3',5'); $\delta_{\text{C}}([^{2}\text{H}_{6}]\text{DMSO})$ 13.2 (CH₃), 27.5 (CH₂), 53.4 (OCH₃), 107.6 (C), 128.3 (2 × CH), 131.7 (C), 132.6 (2 × CH), 134.1 (C), 162.3 (C), 167.1 (C), 169.3 (C); m/z (HRMS-EI) 264.0902 (M⁺ +1). C₁₃H₁₅ClN₃O requires 264.0902; 2,4-diamino-5-[4-chloro-3-(4-methoxyphenyl)phenyl]-6-ethylpyrimidine (10), from **9b** (47%), **9c** (47%) and **9d** (20%), from 50% aqueous ethanol, m.p. 242–244°C; v_{max} (KBr/cm⁻¹) 3453, 3314, 3179, 1630, 1555, 1435, 1238; $\delta_{\rm H}$ ([²H₆]DMSO) 1.00 (3 H, t, J 7.5, CH₃), 2.17 (2 H, q, J 7.5, CH₂), 3.82 (3 H, s, OCH₃), 5.71 (2 H, br s, NH₂), 5.91 (2 H, br s, NH₂), 7.05 (2 H, dt, J 2.5, 8.75, H-2",6"), 7.17 (2 H, m, H-2',6'), 7.44 (2 H, dt, J 2.5, 8.75, H-3",5"), 7.59 (1 H, d, J 8.0, H-5'); $\delta_{\rm C}$ ([²H₆]DMSO) 13.4 (CH₃), 27.7 (CH₂), 55.4 OCH₃), 105.4 (C), 113.8 (2 × CH), 130.2 (C), 130.5 (C), 130.8 (2 × CH), 131.0 (CH), 131.2 (CH), 133.7 (CH), 135.7 (C), 139.9 (C), 159.0 (C), 162.2 (C), 162.4 (C), 166.7 (C); m/z (HRMS-EI) 355.1324 (M⁺ +1). $C_{19}H_{20}CIN_4O$ requires 355.1325.

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References

- 1 Part 36. D.C.M. Chan, C.A. Laughton, S.F. Queener and M.F.G. Stevens, *Bioorg. Med. Chem.* 2002, **10**, 3001.
- N. Miyaura, T. Yanagi and A. Suzuki, *Synth. Commun.* 1981, 11, 513; V.N. Kalinin, *Synthesis*, 1992, 413.
- 3 G.T. Crisp and V. Macolino, Synth. Commun. 1990, 20, 413; S. Gronowitz, A.-B. Hörnfeldt, V. Kristjansson and T. Musil, Chem. Scr. 1986, 26, 305; J. Stavenuiter, M. Hamzink, R. van der Hulst, G. Zomer, G. Westra and E. Kriek, Heterocycles, 1987, 26, 2711; D. Peters, A.-B. Hörnfeldt and S. Gronowitz, J. Heterocycl. Chem. 1990, 27, 2165.
- 4 D.R. Hannah, E.C. Scherer, R.V. Davies, R.B. Titman, C.A. Laughton and M.F. G. Stevens, *Bioorg. Med. Chem.* 2000, 8, 739.
- 5 R.O. Roblin and J.P. English, US Patent, 2,309,739; Chem. Abs. 1943, 37, 3768.
- 6 P.B. Russell and G.H. Hitchins, J. Am. Chem. Soc. 1952, 73, 3763
- 7 E.A. Bliss, R.J. Griffin and M.F.G. Stevens, *J. Chem. Soc.*, *Perkin Trans.1*, 1987, 2217.
- 8 R.J. Griffin, M.A. Meek, C.H. Schwalbe and M.F.G. Stevens, J. Med. Chem. 1989, 32, 2468.
- 9 S. Kano, Y. Yuasa, S. Shibuya and S. Hibino, *Heterocycles*, 1982, 19 1079
- 10 D.J. Brown and B.T. England, J. Chem. Soc.(C), 1971, 425.